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CONSUMPTION OF O₂ IN THE LIGHT BY *CHLORELLA PYRENOIDOSA* AND *CHLAMYDOMONAS REINHARDTII**

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SUMMARY

A mass spectrometer has been used to measure O₂ production and consumption and net CO₂ exchange near the CO₂ compensation point in *Chlorella pyrenoidosa* and *Chlamydomonas reinhardtii* grown in air and in 2% (v/v) and 5% (v/v) CO₂ in air. Some experiments were run in presence of α -hydroxy-2-pyridinemethanesulfonic acid and, for others, data are presented from parallel assays for enzymic glycolate oxidation. Although the glycolate-oxidising enzyme was present in cells grown with and without CO₂ enrichment, the activities measured were not compatible with direct measurements of O₂ consumption in the light. This and other evidence argues against glycolate being the primary substrate for light-induced O₂ consumption by these organisms.

INTRODUCTION

After early and equivocal evidence, BROWN AND WEIS¹ published data from mass spectrometry to show that, in *Ankistrodesmus*, CO₂ evolution was reduced in the light, largely independent of intensity, while O₂ consumption was unaffected at low but increased at high intensities. HOCH AND KOK² made design modifications to their mass spectrometer which made possible the direct sampling of exchange gases without the need for a headspace. With this system, HOCH *et al.*³ examined O₂ exchange in *Anacystis* and in *Scenedesmus* and found O₂ consumption inhibited at low and stimulated at higher intensities. HOCH *et al.*³ suggested that the increased O₂ consumption in the light could be linked with ATP formation.

ARNON *et al.*⁴ and FORTI AND JAGENDORF⁵ have shown a coupling of ATP formation and O₂ uptake with water as electron donor in the light. KRALL AND BASS⁶ observed that O₂ is needed for light-dependent conversion of glucose to starch in leaf discs. The O₂ requirement, however, was low. This was the case also in an O₂ consuming reaction involving photosystem I studied by HEBER AND FRENCH⁷. Although the increased O₂ consumption recorded by HOCH *et al.*³ may reflect ATP formation, the matter is not resolved and it seems possible that at least part of the increased consumption may be traceable to simple back reactions with one or

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more reductants produced in photosynthesis as demonstrated in chloroplasts by MEHLER⁸.

OZBUN *et al.*⁹, working with bean leaves, obtained data which confirmed those of HOCH *et al.*³. In addition, they noted a depression in CO₂ evolution in the light well before photosynthesis was saturated and considered O₂-dependent ATP formation and glyoxylate formation *via* glycolate, without CO₂ release, as alternative mechanisms to explain their observations. GOLDSWORTHY¹⁰ also obtained indications with tobacco that the substrate for CO₂ production changes during photosynthesis. However, he and a number of other investigators including DECKER¹¹, HEW AND KROTKOV¹², JOLLIFFE AND TREGUNNA¹³, MOSS¹⁴, POSKUTA¹⁵, TREGUNNA *et al.*¹⁶ and ZELITCH¹⁷ have presented evidence based on measurements of CO₂ exchange to show that another process, referred to as photorespiration, results in markedly increased CO₂ release in the light in a variety of higher plants. Criteria for photorespiration include a post-illumination decline in CO₂ release to the normal dark rate and an elevation of the CO₂ compensation point with increasing concentrations of O₂. Depression of apparent photosynthesis by O₂ has been attributed to this process, which is favored by low concentrations of CO₂ and high light intensities.

Work, notably by ZELITCH¹⁷ has led to the conclusion that the substrate of photorespiration is glycolic acid. Although glycolic acid oxidase has been demonstrated in a number of higher plants and in some algae including *Chlorella*¹⁸, FOCK AND EGGLE¹⁹ observed that CO₂ formation by this alga in the light was relatively much weaker than in higher plants. VOZNESENSKII²⁰ has reported a zero CO₂ compensation point in *Chlorella pyrenoidosa* and BJORKMAN²¹ found that net CO₂ fixation by *Chlorella* and *Ulva*, unlike higher plants, was not susceptible to O₂ concentration. Using *Solidago* and *Mimulus*, in which net CO₂ fixation is reduced by O₂, BJORKMAN²¹, moreover, was unable to substantiate glycolic acid as the substrate in photorespiration.

It is clear that several alternatives present themselves for the interpretation of data on photosynthesis based on gas exchange²² and that each one may be operative to some extent according to the species under study and environmental conditions. We have undertaken mass spectrometer studies addressed to this problem

MATERIALS AND METHODS

The work was initiated at RIAS*, Baltimore and continued in this laboratory. Methods at RIAS for culturing, harvesting and preparing the experimental organism *Chlamydomonas reinhardtii* have been described²³. Here, it need be noted only that cultures, provided by Dr. B. Kok, were grown at pH 4.2 and gassed with 2% (v/v) CO₂ in air.

Measurements of gas exchange were made at 30° with a mass spectrometer equipped essentially as described by HOCH AND KOK². The method of illumination was identical. In addition to following signals for O₂ of masses 32 and 36, net exchange of CO₂ was also monitored. In all cases, light intensities were adequate for full saturation of photosynthesis although a filter which blocked wavelengths < 460 nm was used routinely. This procedure was found necessary to avoid a very rapid photo-oxidation of the inhibitor α -hydroxy-2-pyridinemethanesulfonic acid which was used in some of the experiments. Use of the filter did not affect the course of O₂ exchange.

* Research Institute for Advanced Study.

From this laboratory, data are presented for two organisms: *Chlorella pyrenoidosa*, strain 211/8P, kindly provided by Dr. M. J. Merrett and *Chlamydomonas reinhardtii*, strain 137c mt⁺ for which we thank Drs. I. Zelitch and P. R. Day. Both species were grown in LEVINE AND EBERSOLD'S²⁴ medium at an incident intensity of 1000 ft candles and gassed either with air or 5% (v/v) CO₂ in air. Incubation temperature was 24°. The mass spectrometer used had similar characteristics to the instrument at RIAS and the sample vessel serving as inlet system was basically identical. Measurements were made at 24°.

RESULTS AND DISCUSSION

Figs. 1 and 2 are from data collected at RIAS and show the time courses of rates of O₂ production and consumption as well as net CO₂ exchange before, during and immediately after illumination with saturating light. In order to monitor changes in rates of CO₂ exchange after illumination, it was necessary to forego monitoring O₂ over the same interval. The results shown in Fig. 1 were obtained with high, and those

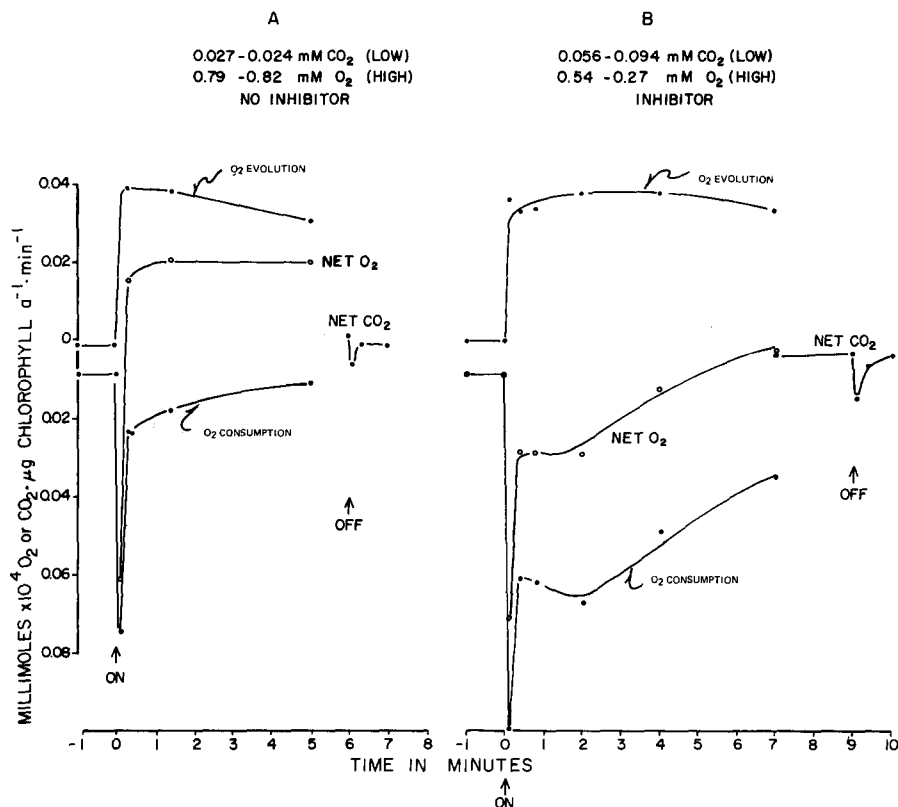


Fig. 1. Time-course of O₂ production and consumption, net O₂ exchange and net CO₂ exchange during a dark, light, dark cycle in *Chlamydomonas*. Low level CO₂ and high level O₂. The inhibitor was α -hydroxy-2-pyridine methanesulfonic acid ($8.8 \cdot 10^{-3}$ M). Note that rates of CO₂ exchange have been inverted to simplify comparison with the data for O₂. Data points were calculated at appropriate points from continuous recorder traces.

in Fig. 2 with low levels of dissolved O₂ relative to water in equilibrium with air at 30° (close to 0.24 mM O₂). Concentrations are shown in the figures together with comparable information for CO₂ which was held at relatively low levels throughout: close to but not reaching the CO₂ compensation point. In each figure, two runs have been included; one without and one with α -hydroxy-2-pyridine methanesulfonic acid which inhibits photorespiration. Ignoring the brief but pronounced "gulps" of O₂ shown in Fig. 1, not yet studied in any detail, we wish to draw attention to the following features. It will be seen that rates of O₂ consumption were higher in the light than in the dark. These rates did not increase with time as the concentration of CO₂ dropped, but declined. Furthermore, addition of the sulfonate resulted in increased rates of O₂ consumption in the light and did not eliminate or reduce the post-illumination decline in CO₂ production as would be expected were this phenomenon attributable solely to glycolate photorespiration. Effects of varying the O₂ concentration were not obvious although these may have been masked or offset by inevitable variations in CO₂ concentration from experiment to experiment. The dramatic reversal in O₂ consumption shown in Fig. 2 in presence of sulfonate is attributed to severe depletion of the reservoir of dissolved O₂ in the reaction vessel with the added complication that O₂ production was halted almost completely with the approach to anaerobic conditions.

While these results argued against the observed O₂ consumption reflecting a glycolate-dependent photorespiration, the possibility remained that glycolate oxidase might have been suppressed by culturing the algae in 2% CO₂. Accordingly, the opportunity was taken to continue the experiments in this laboratory.

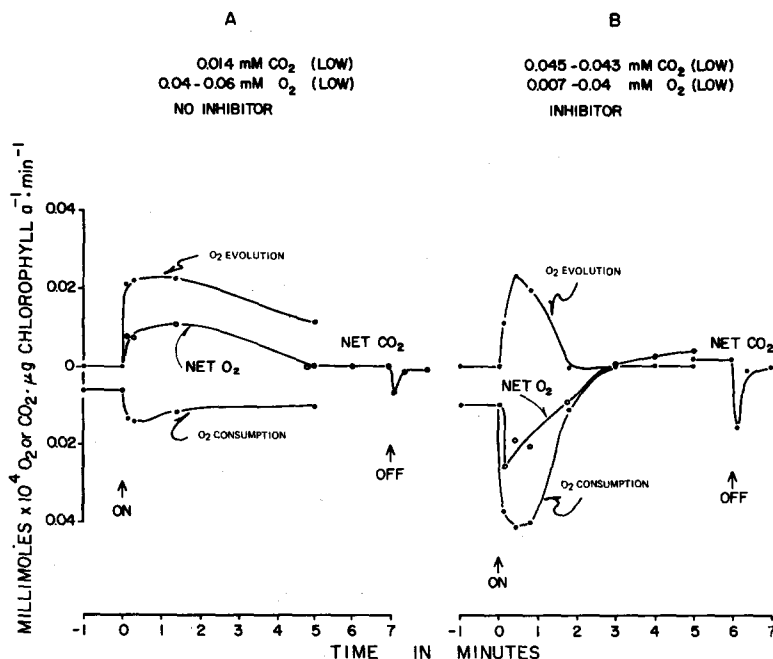


Fig. 2. Time-course of O₂ production and consumption, net O₂ exchange and net CO₂ exchange during a dark, light, dark cycle in *Chlamydomonas*. Low level CO₂ and low level O₂. Other details as in Fig. 1.

TABLE I

O₂ EXCHANGE AT CO₂ COMPENSATION POINT

Culture	$\text{mmoles} \times 10^4 \text{ O}_2 \cdot \mu\text{g chlorophyll } a^{-1} \cdot \text{min}^{-1}$			O ₂ consumption equivalent to glycolate-dichlorophenol- indophenol oxidoreductase* activity
	O ₂ consumption		O ₂ production	
	Dark	Light		
Chlamydomonas				
0.03% CO ₂	0.003	0.020	0.016	0.00030
5% CO ₂	0.003	0.006	0.009	0.00002
Chorella				
2 day culture				
0.03% CO ₂	0.001	0.004	0.005	0.00040
5% CO ₂	0.001	0.006	0.006	0.00012
3 day culture				
0.03% CO ₂	0.001	0.007	0.005	0.00033
5% CO ₂	0.001	0.003	0.003	0.00008

* Cooksey, MS.

Data for observed rates of O₂ exchange by *C. reinhardtii* and *C. pyrenoidosa* at the onset of CO₂ compensation have been set out in Table I. Comparative data have been listed for cells grown in 5% (v/v) CO₂ in air and in air without CO₂ enrichment, in the case of *Chlorella* using cells from 2 day (log phase) and 3 day (late log phase) cultures. Air-grown *Chlamydomonas* was in log phase and the CO₂-grown cells were approaching the stationary phase. Also shown are rates of photorespiratory O₂ consumption that would be predicted from parallel *in vitro* glycolate-oxidising activities obtained by Dr. K. Cooksey using the same material for an associated study (Cooksey, MS).

Notice that O₂ consumption equivalent to glycolate oxidase activity *in vitro* was consistently higher in air-grown than in CO₂-grown cells. Observed O₂ consumption in the light was relatively higher in air-grown *Chlamydomonas* and in the older air-grown *Chlorella* culture. However, there was a wide disparity between predicted and observed O₂ consumption in the light. Furthermore, the ratios of predicted and observed uptakes in each case were in disagreement; in the 2-day cultures of *Chlorella*, where O₂ consumption in the light was slightly higher in CO₂-grown than in air-grown cells, the difference in predicted activity persisted, and, for the air-grown cells, was clearly higher than in *Chlamydomonas*.

Demonstration of glycolate oxidation *in vitro* has been taken as evidence for photorespiration in whole cells or organisms. However, the present study failed to demonstrate agreement between enzyme activity using an artificial electron acceptor and observed O₂ consumption in the light. In fact, CODD *et al.*²⁵ indicate that, in the same algae, O₂ consumption is not associated with the oxidation of glycolate. If this is true, it would follow that the O₂ consumption observed in the light in these experiments is not caused by glycolate oxidation. Whatever the process, it persists in cells grown with high levels of CO₂ and is stimulated, not inhibited, by α -hydroxy-2-pyridine methanesulfonic acid. Other authors (*e.g.* ASADA AND KASAI²⁶) have suggested that sulfonates are not specific for the enzyme responsible for glycolate oxidation.

Finally, it should be noted that the rates of O₂ exchange measured by the mass spectrometer technique are possibly conservative. Although the rate of consumption of both isotopes of O₂ in the dark is proportional to their individual concentrations, it is possible that mass 32 O₂ produced during photosynthesis is preferentially consumed before an equilibrium can be established with isotope of mass 36. Such an effect would serve only to reflect greater rates of O₂ production and consumption than we have reported but would not alter the character of the observations.

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